Clavularins, a New Class of Cytotoxic Compounds isolated from the Soft Coral, *Clavularia koellikeri*

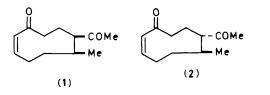
Mamoru Endo,* Masashi Nakagawa, Yoshihiro Hamamoto, and Toshihiro Nakanishi

Suntory Institute for Biomedical Research, Wakayamadai-1 Shimamoto-Cho, Mishima-Gun, Osaka 618, Japan

The structures of two new cytotoxic compounds, clavularins A and B have been elucidated by studies of their ¹H n.m.r. spectra at 360 MHz.

Here we report the structural elucidation of a new class of cytotoxic compounds, \dagger clavularin A (1) and clavularin B (2), isolated from the soft coral, *Clavularia koellikeri* (Thomson and Dean) which lives in the shallow water of a coral reef in Ishigaki-Jima, Okinawa, Japan.

A methanol extract of the soft coral was fractionated on a silica gel column, through a TSK gel column (HW-40 coarse, Toyo Soda Co.), and by silica gel t.l.c.; the fractions were assayed against cultured cells. Two cytotoxic substances were isolated as colourless oils: clavularin A, $M^+ m/z$ 194.1327 (C₁₂H₁₈O₂ requires 194.1306), λ_{max} (n-hexane) 221 nm (ϵ 9 700), v_{max} (film) 2930, 1720, and 1679 cm⁻¹ and clavularin B,



 \dagger Both compounds have a strong lethal effect on PV4 cultured cells transformed with polyoma virus; T/C 50% was observed at 0.25 $\mu g/ml.$

 $M^+ m/z$ 194.1307; λ_{max} (n-hexane) 221 nm (ϵ 10 000), ν_{max} (film) 2920, 1715, and 1675 cm⁻¹.

The skeletal structure of clavularin A was determined by intensive decoupling studies of its ¹H n.m.r. spectra; twodimensional n.m.r. techniques^{1,2} were applied in order to confirm the correlations and assignments.[‡] The *cis*-relation of the olefinic protons was confirmed by nuclear Overhauser enhancement (n.O.e.) (23%) of 2-H (δ 5.97 in C₆D₆) caused by irradiation at 3-H (δ 6.27 in C₆D₆).

[‡] N.m.r. data for clavularin A: ¹H n.m.r. (CDCl₃) δ 6.76 (1H, ddd, J 11.7, 7.2, and 4.1 Hz, 3-H), 6.02 (1H, d, J 11.7 and 2.7 Hz, 2-H), 2.81 (1H, ddd, J 9.7, 5.4, and 4.1 Hz, 7α-H), 2.49 (1H, ddd, J 17.1, 9.0, and 5.4 Hz, 9β-H), 2.42 (2H, m, 4α-H and 4β-H), 2.31 (1H, ddd, J 17.1, 8.2, and 6.6 Hz, 9α-H), 2.12 (3H, s, 7β-COMe), *ca*. 2.1 (3H, m, 5α-H, 6α-H, and 8α-H), 1.61 (1H, m, 8β-H), 1.26 (1H, m, 5β-H), and 0.83 (3H, d, J 7.2 Hz, 6β-Me); ¹³C n.m.r. (CDCl₃) δ 208.2 (s), 203.2 (s), 148.1 (d), 133.5 (d), 53.7 (d), 41.5 (t), 35.4 (t), 35.5 (d), 29.5 (q), 27.4 (t), 22.1 (t), and 15.7 (q) p.p.m. N.m.r. data for clavularin B: ¹H n.m.r. (CDCl₃) δ 6.55 (1H, ddd, J 11.5, 6.2, and 3.5 Hz), 5.95 (1H, ddd, J 11.5, 2.4, and 0.9 Hz), 2.51–2.27 (5H, m), 2.11 (3H, s), 1.98–1.60 (5H, m), and 1.09 (3H, d, J 6.8 Hz); ¹³C n.m.r. (CDCl₃) δ 208.3 (s), 205.9 (s), 145.5 (d), 131.8 (d), 59.3 (d), 41.5 (t), 34.6 (d), 33.3 (t), 29.9 (q), 27.7 (t), 24.6 (t), and 19.9 (q) p.p.m.

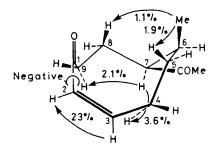


Figure 1. N.O.e.s, H (irradiated) \rightarrow H (enhanced), and the negative skew of the α,β -unsaturated ketone in clavularin A.

Owing to a shielding effect of the carbonyl group, the ¹H n.m.r. signal of 5 β -H in clavularin A appears at unusually high field (δ 1.26 in CDCl₃ and 0.98 in C₆D₆). The conformation from C-1 to C-6 was thus elucidated (Figure 1) and the existence of a skewed α,β -unsaturated ketone was proved by c.d. spectroscopy (see below). The β -configuration of 6-Me was indicated by n.O.e. (1.9%) of the 5 β -H ¹H n.m.r. signal (δ 1.26 in CDCl₃) caused by irradiation at 6-Me (δ 0.83 in $CDCl_3$). The α -configuration of 7-H was indicated by n.O.e.s of one 4-H (3.6 %, δ 2.42 in CDCl₃) and one 9-H (2.1 %, δ 2.31 in $CDCl_3$) caused by irradiation at 7-H (δ 2.81 in $CDCl_3$) (7-H can be located in the vicinity of 4-H and 9-H only when it is in the α -configuration and when C-7, C-8, and C-9 are in the conformation shown in Figure 1). The 4-H and 9-H which show n.O.e.s were thus assigned as 4α -H and 9α -H. The n.O.e. difference spectrum technique³ was applied throughout the n.O.e. studies.

323 A was based on the c.d.

The absolute structure of clavularin A was based on the c.d. Cotton effect;^{4,5} a negative sign ($\Delta \epsilon - 7.08$) for the K-band (224 nm) and a positive sign ($\Delta \epsilon + 3.89$) for the R-band (339 nm) are due to the negative skew of the α,β -unsaturated ketone (Figure 1).

N.O.e. (3.6%) of 7-H (δ 2.21 in C₆D₆) in clavularin B caused by irradiation at 6-Me (δ 0.83 in C₆D₆) suggests that clavularin B is the 7-epimer of clavularin A. This was confirmed by the conversion of clavularin A into clauvularin B by hydrogen chloride in methanol.

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